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قياس الأطوال الموجية والزمن الطيفي لمركبات فسفورية عالية الإصدار الضوئي

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🗆 الملخّص 🗆

نظراً للاهتمام الحالي الكبير للمركبات المتفسفرة عالية الإصدار الضوئي في مجال الأطوال الموجية الكبيرة (550-1050nm) ولتطبيقاتها الواسعة في مختلف المجالات الفيزيائية (فيزياء اللايزر-فيزياء أنصاف النواقل-الفيزياء الطبية والتصوير المقطعي الطبقي....الخ).

ونظراً لكون طيف الإصدار الضوئي لهذه المركبات متوافقاً مع حساسية الثنائيات الضوئية من السيليكون، فقد بينا في هذه المقالة مميزات زمن التضاؤل والزمن الطيفي لبعض المركبات المتفسفرة ذات الأطوال الموجية الطويلة مثل:AL2O3:Ti, A^PB ^{VI} وبعض الموليبدات، Y2O2S و Gd2O2S مع Tb ومواد أخرى. كذلك تم دراسة انتقال سويات الطاقة نحو مراكز التألق وعلاقتها بالانتقالات النسبية.

أجريت القياسات في جامعة بطرس بورغ التكنولوجية الحكومية – بطرس بورغ– روسيا .

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INTRODUCTION

Scintillator applications increase from year to year. One of the best photodetectors for emitting light is a silicon photodiode. In this paper we report our results in search of new phosphors. This search covers long emission wavelengths because applications of silicon photodiodes would benefit from scintillators that emit in the 550-1050 nm wavelength region. We consider the following requirements to studied materials: high light output, short decay time, chemical, mechanical and radiation resistance, and low level of afterglow. Very often we prefer one of the characteristics if it was necessary to solve a concrete task.

EXPERIMENTAL PROCEDURES

The emission spectra have been measured at steady state X-ray excitation (45kV). Two photomultiply tubes (PMT) in different wavelength regions were used. In a short wavelength region it was PMT- 106 (200-680nm) and in a long wavelength region it was PMT- 83 (400-1100nm). In some cases PMT-106 was more preferred than PMT-83, because in wavelength region 400-650 nm PMT -106 is more sensitive. Monochromator MDR-2 was used with both PMTs with gratings 1200splits/mm for PMT-106 and 600 splits/mm for PMT-83. The temperature was varied in 80-500 K range. All the spectra were compared with spectral sensitivity of silicone photodiode. [1] All kinetic measurements have been made using pulsed (less then 1 ns) X-ray source and detection START-STOP standard system. The description of this setup was published by Rodnyi et al. [2] To determine relative light output two methods were used. In both cases standard sample was Csl:Tl to obtain this data in the first method we compared the areas under the spectrum curves of a measured sample and a standard sample. In the second one the number of counts in 50 µs region for a measured sample and a standard one were compared. For level of afterglow measurements we used ultraviolet (UV) excitation (Hg-lamp) and the device for kinetic measurements. To determine the trap properties such as energy levels, their relative concentrations, we used thermo-stimulated luminescence (TSL) method. The TSL measurements were carried out after UV (exposure 2 min) or X-ray (exposure 10 min) excitations. The exposures were carried out at 295 K and glow curves were measured at heating up to 600 k with constant rate (the rate can be varied from 0.5 to 1.0 k/s.) see Fig 1

RESULTS AND DISCUSSION

1- Tikor

monocrystals of Al₂ O₃: Ti (tikor) is mainly used as active component for powerful tunable lasers. And now tikor attracts attention as scintillation material [3]. The samples were grown by method of horizontal oriented crystallization in vacuum and argon-carbon atmosphere with velocity 1.5-6 mm/h. the proportion of absorption coefficients in wavelength region 490 nm and 800nm, which characterizes Ti³⁺ /Ti⁴⁺ proportion, was equal 200 for AL₂O₃: Ti (0.07mass %) sample. At room temperature we observe three main wide bands: long –wavelength band at 705 nm with 0.31 eV halfwidth, short wavelength band at 315 nm with 0.8 eV halfwidth, and not intensive one at 420 nm. The intensive long –wavelength luminescence with decay time constant 3.4 ± 0.1 µs associated withTi³⁺ transmission ²E ® ²T₂ in AL₂O₃ [4,5]. The intensity of 705 nm band

increases (about 50%) with the temperature decreasing to 80 k. the relative intensity of this band is equal to 14% of standard scintillator Csl : Tl at room temperature. The short wavelength band shows two decay time constants: $t_1 = 0.13 \pm 0.01$ ms and $t_2 = 1.3 \pm 0.1$. ms. Its intensity is growing more than three times with temperature decreasing. The band at 420 nm is usually associated with charge transfer emission [5,6]. In this case Ti⁴⁺ traps an electron and becomes Ti³⁺ with a quantum emission. In introduction we paid attention to possibility of using long-wavelength scintillator with silicon photodiodes as a photodetector. Fig 2 presents the silicon photodiodes sensitivity [1] compared to the tikor spectrum.

2- Y and Gd oxysulphides doped with Tb

Yttrium and gadolinium oxysulphides doped with rare earth elements are widely used as efficient cathode-and X-ray phosphors [7,8]. We studied ceramic Y_2O_2S : Tb and Gd_2O_2S :Tb samples. The spectra of this samples contained only lines associated with radiative transitions in the activators. We observed ${}^5D_3{}^{-7}F_j$ and ${}^5D_4{}^{-7}F_j$ (J= 1-6) transitions in the emission spectra (see Fig 3) This figure shows the transition lines spectrum at definite wave length for the Tb doped in the Ceramic. compound we have studied The kinetic measurements show decay time constants in range of 150-250 ms. In this investigation we are primarily interested in the light yield output of this samples in comparison with CsI:Tl. We were interested in light yield output for photons which had the same direction with incident X-ray and for photons which were emitted at the sample side opposite to incident X-ray. The other point of our interest was such characteristic as afterglow. We separated the samples with smallest level of afterglow and highest light output. Thus technological conditions of the ceramic preparations have determined. For more clear understanding of technological condition influence on luminescence properties of the samples the TSL measurements have been also fulfilled.

3- Molybdates

Different tungstates are known as good scintillators with high density, while a little known about scintillation properties of Molybdates. Very often Molybdates are followed with tungstates. Apart from the $WO_3^{2^-}$ group the Mo impurity (i.e. the (MoO₄) ²⁻ group) was ascribed to the emission component around 520 nm[9]. We studied some molybdate crystals such as CaMoO₄, SrMoO₄, NaLa (MoO₄)₂. The crystals exhibit broad (~ 0.7 eV) emission bands peaked at 537 nm (CaMoO₄), 513 nm (SrMoO₄), 540 nm and 605 nm (LaNa (MoO₄)₂). Kinetic measurements show that these scintillation materials have two exponential components, one of which is fast. We obtained the following data of decay time constant for this scintillators (fast/slow components): 90 ns /5.0 ms, 4.2 ns /0.9 ms and 4.2 ns/1.2 ms for CaMoO₄, SrMoO₄ and NaLa(MoO₄)₂, respectively. The SrMoO₄ exhibits largest light output among the crystals. A little is known about luminescence centres in molybdates. We can suppose that the nature of luminescence is self-trapped exitons (STE) and that the fast and slow emission components are related to singlet and triplet STE's correspondingly.

4- $A^2 B^6$:

 $A^2 B^6$ compounds are coming into use as effective phosphors. But there are some problems to grow single crystal phosphor on base of ZnS. We studied scintillation properties of transparent homogeneous polycrystalline samples of Zns and ZnS-ZnSe solid solutions doped by Ag and Cu. We investigated spectral characteristics, which varied from sample to sample depending on growing conditions. For example ZnS: Cu samples show from one (534 nm) to three (456, 518, 608 nm) peaks depending on condensation temperature and barium concentration in flax. In ZnS-ZnSe solid solution there is no so high dependence but we observed a high difference in luminescence intensity for this samples. At this moment we can't explain the kinetic properties of luminescence centres of this samples because many influences on it (temperature conditions of the samples preparation, type of flux and its concentration, concentration of activators and so- activators, correlation between wurzite and sphalerite phase in the samples). We can only say that the shortest decay was 13 ns in a sample of this set of compounds. All resuls are summerized in table 1.

Sample	l _m , nm	t, ns
Al_2O_3 : Ti	315	130
	420	1300
	705	3400
Gd ₂ O ₂ S: Tb	370-730	$1.5 - 2.5 \cdot 10^5$
CaMoO ₄	537	90,5000
SrMoO ₄	513	4.2,900
$LaNa(MaO_4)_2$	540,605	4.2,1200
ZnS-ZnSe(50%50%): Cu,Al	580	13
Na(Nd) MgAl (Cr) ₁₁ O ₁₉	694,709,750	453
CdI ₂	532,564	2.4
BaSO ₄ :Pr	480,600,637	9

Table1:Spectral and kinetic properties of some long-wavelength phosphors.

CONCLUSIONS

The findings show that the characteristics of some studied scitillators can be improved. Therefore, we have to choose optimal growth conditions, concentration of activator and co-activator, and type of flux. Besides, we plan to vary the post-growth conditions, particularly the cooling rate

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Fig 1: shows the instruments used in measuring the RL (X-Ray luminescence), TSL (Thermoluminescence)

1-Variable frequency generator.	2- Rheostat.	
3- X-Ray pulses generator.	4- High power supply.	
5- Control unit.	6- Modulating power supply.	
7-Variable power supply.	8-Cut universal.	
9-Time control connected to channel CTRAT.	10-Variable time modulator.	
11- Frequency measuring.	12- Time control connected to channel CTOP.	
13- Amplifier.	14-Optical electron generator.	
15-Power supply for optical electron generator.	16- Variable digital analyzers.	
17- IBM PC computer.	18-Cryostat.	
19-Windows for cryostat Of Beryllium.	20-Quarts windows.	
21- Sample.	22- Analyzer.	



Fig 2. Normalized silicon photodiodes sensitivity (1) and tikor emission spectrum (2)



Fig 3. ENERGY LEVELS OF THE +3 LANTHANIDES IN LaF3

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The Octupole Correlations In The ²⁰⁹bi,²⁰⁹pb Nuclei

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\Box ABSTRACT \Box

The self consistent Hartree-Fock-Bogoliubov problem for nuclei with A=209 is solved by using the variation principle. Accordingly, the octupole coupling of $h_{\frac{9}{2}} \otimes d_{\frac{3}{2}}$ and $i_{\frac{13}{2}} \otimes h_{\frac{9}{2}}$ of the nucleus ²⁰⁹Bi have been calculated for different octupole strengths. Moreover, the octupole coupling between $j_{\frac{15}{2}} \otimes g_{\frac{9}{2}}$ of the nucleus ²⁰⁹Pb has

been also calculated.

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$^{209}{ m Bi},\,^{209}{ m Pb}$ ارتباطات ثماني القطب في النواتين

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(قبل للنشر في 2003/7/19 (

🗆 الملخّص 🗆

حلت مسألة هارتري -فوك -بوغليوبوف للأنوية التي لها عدد كتلي A=209 باستخدام مبدأ التغير. وتم حساب ارتباط ثماني قطب للإنتقالين $\frac{1}{2}$ B $h_{\frac{9}{2}}$ ، $h_{\frac{9}{2}}$ B $h_{\frac{1}{2}}$ من أجل سعات مختلفة لاهتزاز $\frac{1}{2}$ B $h_{\frac{1}{2}}$ ، $h_{\frac{9}{2}}$ B $h_{\frac{1}{2}}$ من أجل سعات مختلفة لاهتزاز مساب ارتباط ثماني قطب للإنتقال $\frac{1}{2}$ B $\frac{$

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